

Development and Validation of First Order Derivative Spectrophotometric Method for Simultaneous Estimation of Bromhexine Hydrochloride and Phenylephrine Hydrochloride in their Combined Pharmaceutical Dosage Form

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ABSTRACT

UV Spectrophotometric method has been developed for simultaneous estimation of Bromhexine HCl (BHX) and Phenylephrine HCl (PLE) in bulk drug and in their combined Pharmaceutical dosage form by first order derivative. This method utilizes methanol as a common solvent and λ_{max} of BHX and PLE selected for analysis was found to be at 241 nm (at ZCP of PLE) and 233 nm (at ZCP of BHX) respectively. Linearity was observed in the concentration range of 5-30 $\mu\text{g/ml}$ for BHX ($r^2 = 0.999$) and 10-60 $\mu\text{g/ml}$ for PLE ($r^2 = 0.998$). The accuracy and precision were determined and found to comply with ICH guidelines. This method shows good reproducibility and recovery with % RSD in the desired range. Developed method was applied for marketed formulation. The results were found to be within acceptance criteria according to ICH guideline. This method was simple, rapid, accurate, Precise and sensitive.

Keywords: Bromhexine, Phenylephrine, First Order Derivative Spectrophotometric Method

INTRODUCTION ^[1-4]

Bromhexine Hydrochloride (BHX) (fig. 1a) ^[1] is the hydrochloride salt of bromhexine, chemically known as, 2,4-dibromo-6 {[cyclohexyl(methyl) amino]methyl}aniline ^[1]. It act as Mucolytic. It is used in the treatment of respiratory disorder associated with productive cough & abnormal mucus secretion ^[3]. Phenylephrine HCl (PLE) (fig.1b) ^[1] is Sympathomimetics class drug used in the treatment of Nasal decongestant, Vasoconstrictor Agents, Mydriatics & Cardiotonic Agents ^[3]. But here it's used as Nasal decongestant. It's chemically known as 3-[(1R)-1hydroxy-2-(methylamino)ethyl]phenol. ^[1]

BHX is official in IP & BP ^[6,8] and PLE is official in IP, BP, & USP. ^[6,7,8] The chemical structures of BHX & PLE are shown in Fig. 1. Combination drug products of BHX and PLE are widely marketed under brand name (solvin) ^[9] and used in the treatment of cold & cough specially during some allergic condition. Several analytical methods reported such as UV Spectrophotometry, HPLC, HPTLC & Stability study for the estimation of BHX and PLE Individually and its combination with other drug ^[16-19]. But still no one any analytical method developed for the simultaneous estimation of these two drugs. So, the present paper describes the Spectrophotometric method development and validation for the

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simultaneous estimation of BHX & PLE by first order derivative spectrophotometric method in tablet dosage form. The proposed methods are optimized and validated as per the ICH guidelines.

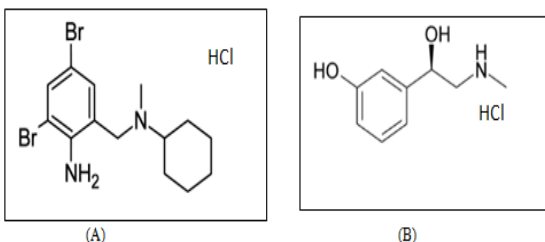


Fig. 1 Chemical structures of the analytes (A) Bromhexine HCl (B) Phenylephrine HCl

MATERIALS AND METHODS ^[10-14]

Instrumentation

Double beam UV-visible spectrophotometer (helios Alpha, Model - V 7.09) having two matched quartz cells with 1 cm light path. An Electronic analytical balance (Contech, CA34 Model) was used in the study.

Material and reagent

Double distilled water and Whatmann filter paper (0.45 μ m) were used for filtration. Active pharmaceutical ingredient (API) Bromhexine HCl, Phenylephrine HCl were obtained as gift sample from Espee formulation Pvt. Ltd, Rajkot and Manish Pharma Lab, Viramgam respectively. And test samples (tablets with composition BHX-8 mg and PLE-10 mg (brand name: solvin) were procured from the local market.

Preparation of Standard Stock solution of BHX and PLE:

Accurately weighed quantity 100 mg of BHX and PLE were transferred into separate 100 ml volumetric flask, dissolved and diluted up to mark with methanol (100 ml). This will give a stock solution having strength of 1000 μ g/ml of each.

Preparation of Working Standard Solution of BHX and PLE:

100 μ g/ml of BHX and PLE solution were prepared by diluting 10 ml of stock solution to 100 ml with methanol in separate 100 ml volumetric flask. Suitable aliquots of this solution were diluted up to the mark with methanol to get the concentration range of 5,10,15,20,25, & 30 μ g/ml for BHX and 10,20,30,40,50, & 60 μ g/ml for PLE.

Selection of analytical wavelength:

5-30 μ g/ml solutions of BHX and 10-60 μ g/ml solutions of PLE were prepared in methanol by appropriate dilution of working standard solution and spectrum was recorded between 200-400 nm and all zero order spectrums (D^0) were converted to first derivative spectrum (D^1) using delta lambda 1.0 and scaling factor 5.0. The overlain first derivative spectrums of BHX and PLE at different concentration were recorded. The zero crossing point (ZCP) of BHX was found to be at 233 nm (Figure 2) and ZCP of PLE was found to be at 241 nm (Figure 3).

Preparation of calibration curve:

Standard solutions of BHX in the concentration range of 5 to 30 μ g/ml obtained by transferring (0.5, 1.0, 1.5, 2, 2.5, & 3 ml) of BHX working standard solution (100 μ g/ml) to the series of 10 ml volumetric flasks and standard solutions of PLE in the concentration range of 10 to 60 μ g/ml were obtained by transferring (1, 2, 3, 4, 5, & 6 ml) of PLE working standard solution (100 μ g/ml) to the series of 10 ml volumetric flasks. Then volume was adjusted up-to mark with methanol. All dilutions were scanned in wavelength range of 200 nm to 400 nm. All zero order spectrums (D^0) were converted to first derivative spectrum (D^1). The absorbance was plotted against the respective concentrations to obtain the calibration curves.

Validation of proposed method ^[15]

Parameters to be considered for the validation of method are:

Validation of the developed method was carried out as per ICH guideline. Parameters such as Linearity and range, Accuracy, Precision, LOD and LOQ were taken up as tests for analytical method validation.

Linearity and Range:

Appropriate volume of aliquot from BHX and PLE working standard solution was transferred to volumetric flask of 10 ml capacity. The volume was adjusted to the mark with methanol to give a solution containing 5-30 µg/ml BHX and 10-60 µg/ml PLE. All D¹ Spectrum were recorded using above spectrophotometric condition. D¹ absorbance at 241 nm and 233 nm were recorded for BHX and PLE, respectively (n=3). Calibration curves were constructed by plotting average absorbance versus concentrations for both drugs. Straight line equations were obtained from these calibration curves.

Precision

Precision of the method was determined in the terms of Repeatability, Intraday and Interday precision. Repeatability (% RSD) was assessed by analyzing test drug solution within the calibration range, six times on the same day. Intraday variation (% RSD) was determined by analysis of this solution three times on the same day. Interday precision (%RSD) was determined by analysis of this solution on three different days.

Limit of detection (LOD) and limit of quantitation (LOQ)

$$\text{LOD} = \frac{3.3 * \text{SD}}{\text{Slope}}$$

$$\text{LOQ} = \frac{10 * \text{SD}}{\text{Slope}}$$

Where,

S.D. = Standard deviation of the Y- intercepts of the 3 calibration curves

Slope = Mean slope of the 3 calibration curves

Recovery Studies:

Recovery studies were done so as to check the accuracy of the method. Known amounts of standard solutions of BHX and PLE were added to pre-quantify dosage form solution of BHX and PLE and D¹absorbance were determined at 241 nm and 233 nm respectively. Concentration of the drug in the mixture was calculated using the regression equations. The analysis was done in a set of 3 replicates. The amount of BHX and PLE was calculated at each level (80%, 100%, 120%) and % recoveries were computed.

Application of Proposed Method to Marketed formulation:

First Order Derivative Spectrophotometry^[12]

The powder of twenty tablets was weighed. An accurately weighed quantity of the powder equivalent to about 8 mg of BHX and 10 mg PLE of was taken in 100 ml volumetric flask and dissolved with methanol and further diluted upto the mark with same solvent. The solution was then filtered through the Whatmann filter paper No. 41. Necessary dilutions are made with methanol to give final concentration 8 µg/ml and 10 µg/ml of BHX and PLE respectively. The solutions are then scanned between 200-400nm and absorbances are measured at respective wavelengths. The concentration of each drug was calculated using equation of straight line.

RESULTS AND DISCUSSION

Zero cross point (ZCP):

ZCP of BHX and PLE were detected at 233 nm and 241 nm, respectively and overlain D¹spectra were recorded (figure 4).Regression characteristics for BHX and PLE are shown in Table 4.

Method Validation:

The linearity range for BHX and PLE were 5-30 $\mu\text{g/mL}$ and 10-60 $\mu\text{g/mL}$ respectively. Recovery studies was carried out by addition of standard drug solution to pre-analyzed dosage form solution at four different concentration levels (0%, 80%, 100% and 120%) taking into consideration percentage purity of added bulk drug sample. The results of the recovery studies are found to be satisfactory for BHX and PLE and shown in Table 1 and 2 respectively. The result of assay procedure obtained was showed in Table 3. Summary of Other validation

parameters including Repeatability, Intraday, Interday, LOD and LOQ were found to be satisfactory and are shown in Table 5.

CONCLUSION

The results obtained by applying the suggested procedures, it is proved that the proposed method is accurate, precise, simple, sensitive, selective and rapid and can be applied successfully in routine analysis for the estimation of BHX and PLE in their combined pharmaceutical dosage form. The developed method was validated as par ICH guidelines.

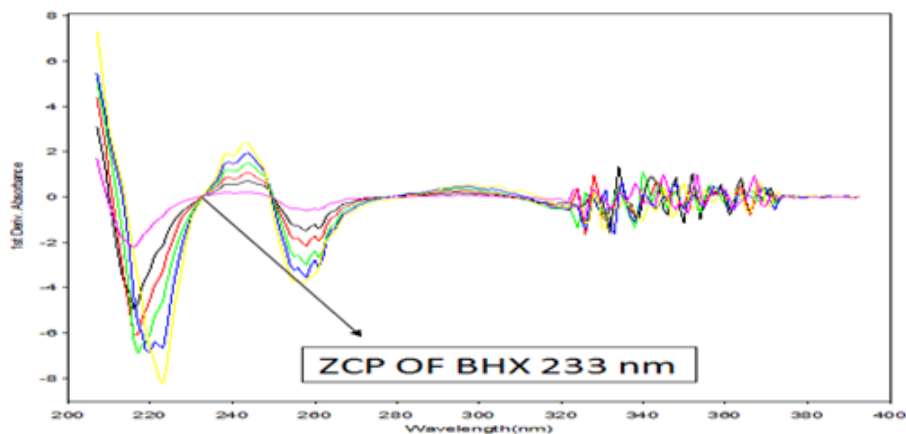
FIGURES AND TABLES

Figure-2: Overlain D^1 Spectrum of Bromhexine HCl (5-30 $\mu\text{g/ml}$) in methanol

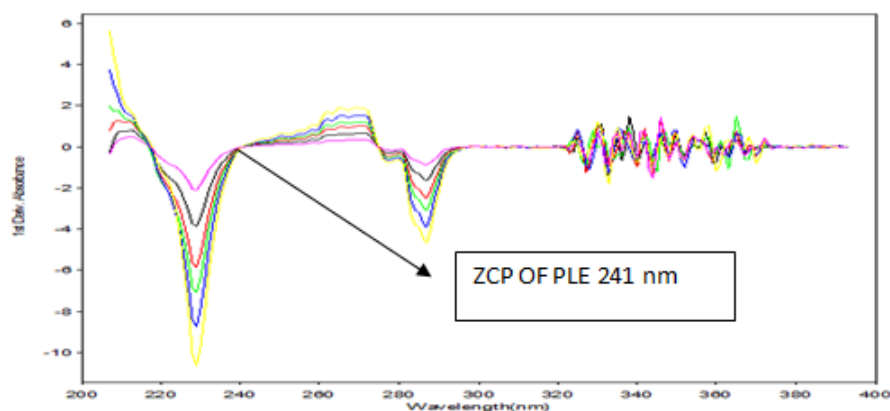


Figure-3: Overlain D^1 Spectrum of Phenylephrine HCl (10-60 $\mu\text{g/ml}$) in methanol.

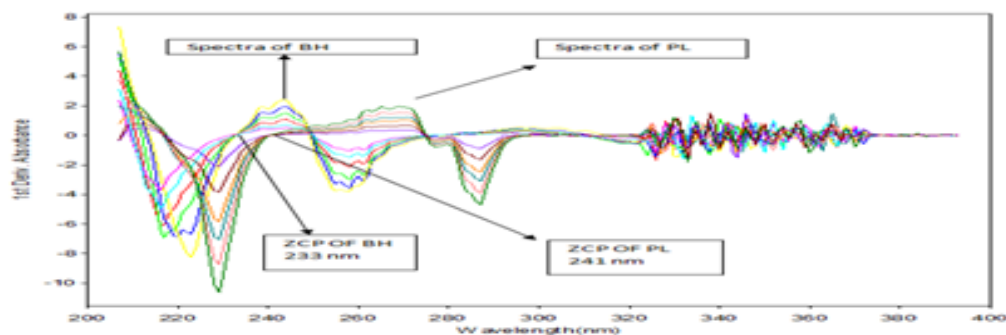
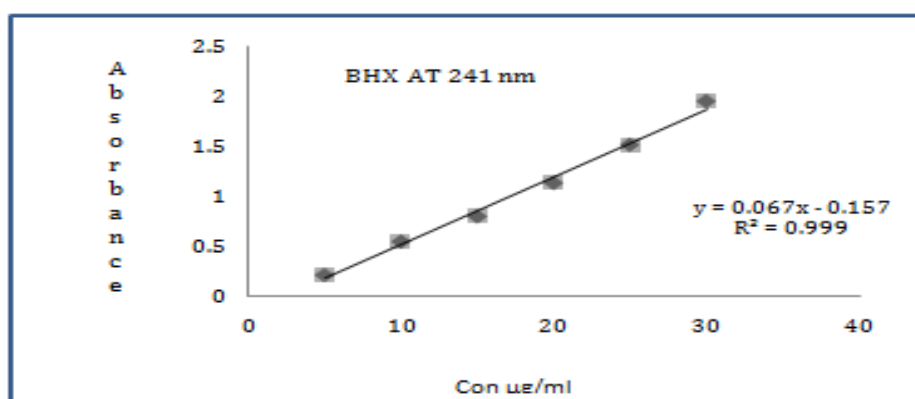
Figure-4: Overlain D² Spectrum of BHX and PLE in methanol

Figure-5: Calibration curve of BHX

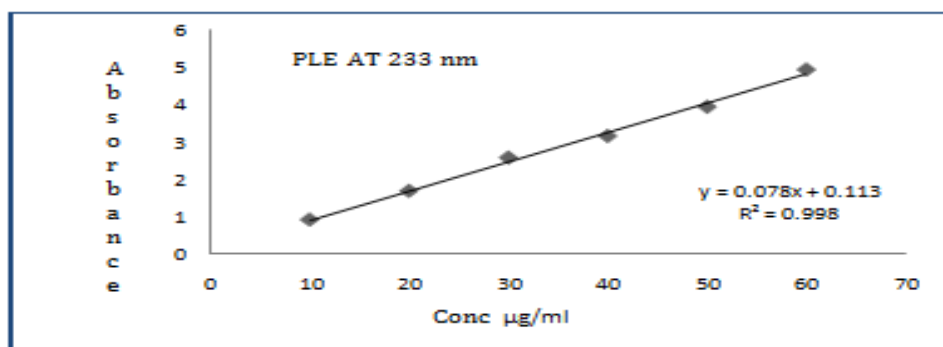


Figure-6: Calibration curve of PLE

Table-1 Accuracy data of BHX in dosage form:

Amount of BHX in Sample (µg/ml)	Amount of std. BHX Added (µg/ml)	Total Amount of BHX (µg/ml)	Mean Total Amount of BHX recovered (µg/ml) Mean ± S.D.*	% Recovery	% R.S.D.
10	0	10	10.17 ± 0.0057	101.7	0.0567
10	8	18	18.27 ± 0.0251	101.5	0.1376
10	10	20	19.62 ± 0.0152	98.1	0.1279
10	12	22	23.43 ± 0.02	101.81	0.0677

[*=mean value of 3 determination]

Table-2 Accuracy data of PLE in dosage form:

Amount of PLE in Sample ($\mu\text{g/ml}$)	Amount of std. PLE Added ($\mu\text{g/ml}$)	Total Amount of PLE ($\mu\text{g/ml}$)	Mean Total Amount of PLE recovered ($\mu\text{g/ml}$) Mean \pm S.D.*	% Recovery	% R.S.D.
20	0	20	20.35 \pm 0.0305	101.75	0.1501
20	16	36	35.84 \pm 0.0360	99.55	0.1006
20	20	40	39.42 \pm 0.0251	98.55	0.0638
20	24	44	44.49 \pm 0.0208	101.11	0.0467

[*=mean value of 3 determination]

Table-3: Analysis of BHX and PLE in dosage form:

Tablet dosage form	Label claim (mg)		% Recovery \pm SD (% of label claim*)	
	BHX	PLE	BHX	PLE
Solvin	8 mg	10 mg	100.25 \pm 1.9423	100.17 \pm 1.6590

[*=mean value of 3 determination]

Table-4: Regression Characteristics:

Characteristics	Bromhexine HCl (at ZCP= 241 nm)	Phenylephrine HCl (at ZCP=233 nm)
Range ($\mu\text{g/ml}$)	5-35	10-100
Linearity ($\mu\text{g/ml}$)	5-30	10-60
Molar Absorptivity (lit. $\text{mol}^{-1} \text{cm}^{-1}$)	2417.76	1695.19
Sandell's sensitivity ($\mu\text{g/cm}^2/0.001$ abs unit)	0.1556	0.1204
Regression Equation	$y = 0.067x - 0.157$	$y = 0.078x + 0.113$
Slope	0.067	0.078
r^2	0.999	0.998
Intercept	0.157	0.113
S.D. of Intercept	0.000816	0.002449

TABLE-5: VALIDATION PARAMETERS:

PARAMETERS	Bromhexine HCl (at ZCP of PLE=241 nm)	Phenylephrine HCl (at ZCP of BHX=233 nm)
Linearity range ($\mu\text{g/ml}$)	5-30	10-60
Precision (% RSD)		
Repeatability (%RSD) (n=6)	0.1014	0.02932
Intra-day (n=3)	0.1213-0.4478	0.096-0.161
Inter-day (n=3)	1.0346-1.9517	0.4901-0.5506
LOD ($\mu\text{g/ml}$)	0.0015	0.002
LOQ ($\mu\text{g/ml}$)	0.2330	0.2858
Accuracy (% recovery)	99.50-100.89	99.37-101.01

LOD: Limit of Detection, LOQ: Limit of Quantitation, R.S.D.: Relative standard deviation, S.D.: Standard deviation

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